# E upha-7,9(11),24-trien-3 $\beta$-ol ("Antiquol C") and Other Triterpenes from Euphorbia antiquorum Latex and Their Inhibitory Effects on Epstein-Barr Virus Activation 

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#### Abstract

The structures of three triterpene alcohols isolated from the latex of Euphorbia antiquorum were established to be eupha-7,9(11),24-trien-3 $\beta$-ol (2; antiquol C), 19(10 $\rightarrow 9$ )abeo-8 $\alpha, 9 \beta, 10 \alpha$-eupha-5,24-dien$3 \beta$-ol (3; antiquol B), and 24-methyltirucalla-8,24(241)-dien-3 $\beta$-ol (4; euphorbol) on the basis of spectroscopic methods. Compounds $\mathbf{3}$ and $\mathbf{4}$ have previously been assigned the erroneous structures of $10 \alpha$-cucurbita-5,24-dien-3 $\alpha$-ol and 24-methyleupha-8,24(24ㄹ)-dien-3 $\beta$-ol, respectively. Compounds 2-4 and four other known compounds isolated from the latex, euphol (1), Iemmaphylla-7,21-dien-3 $\beta$-ol (5), isohelianol (6), and camelliol C (7), showed potent inhibitory effects on Epstein-Barr virus early antigen (EBV-EA) activation induced by the tumor promoter 12-O-tetradecanoylphorbol-13-acetate (TPA).


Euphorbia antiquorum L. (E uphorbiaceae), a plant native to India and Sri Lanka, has various medicinal uses. Its whole plant is used for cutaneous infections, and Iatex for dropsy, as nerve tonic, and for bronchitis. ${ }^{1}$ The latex has been reported to contain euphol (1), euphol 3-Ocinnamate, antiquol $A$, antiquol $B$, 24-methylenecycloartanol, and cycloeucalenol as the triterpene constituents. ${ }^{2}$ In this paper, we report the isolation and characterization of one novel triterpene al cohol, named antiquol C (2), and six known triterpene al cohols ( $\mathbf{1}, \mathbf{3}-\mathbf{7}$ ) from the latex of E . antiquorum, as well as their inhibitory effects on EBV-EA activation induced by TPA, as a primary screening for antitumor promoters. In addition, the structural revision of antiquol B (3) and euphorbol (4) as 19(10 $\rightarrow 9$ )abeo$8 \alpha, 9 \beta, 10 \alpha$-eupha-5,24-dien-3 $\beta$-ol and 24 -methyltirucalla-8,24(24¹)-dien-3 $\beta$-ol, which were previously assigned erroneously as $10 \alpha$-cucurbita-5,24-dien-3 $\alpha$-ol ${ }^{2}$ and 24-methyl-eupha-8,24(24$)$-dien- $3 \beta$-ol, ${ }^{3}$ respectively, is described.

## Results and Discussion

The seven triterpene alcohols were isolated from the latex as their acetyl derivatives, 1a-7a, as described in the Experimental Section, and their identification was confirmed by comparison of chromatographic and spectral data with those of authentic compounds.

The molecular formula of 2 a was determined as $\mathrm{C}_{32} \mathrm{H}_{50} \mathrm{O}_{2}$ from its HREIMS ([M] ${ }^{+}, \mathrm{m} / \mathrm{z} 466.3842$ ) as well as from its ${ }^{13} \mathrm{C}$ NMR DEPT. The compound had a $\Delta^{7,9(11)}$-conjugated diene [ $\lambda_{\max } 230$ ( $\log \epsilon 4.07$ ), 237 (4.09), and 246 (3.99) nm; $v_{\max } 1650,816 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 5.20(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=3.5 \mathrm{~Hz})$ and 5.33

[^0](1H, dd, J = 2.1, 3.1 Hz)], ${ }^{4,5}$ a terminal isopropylidene group [ $\delta_{H} 1.61$ (s) and $1.69(\mathrm{~s})$ ], a secondary [ $\delta_{H} 0.86$ (d)] and five tertiary methyls [ $\delta_{H} 0.62,0.84,0.87,0.94$, and 0.96$]$, and an acetoxyl group [ $v_{\text {max }} 1738 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 2.06(3 \mathrm{H}, \mathrm{s})$ and 4.51 ( $1 \mathrm{H}, \mathrm{dd}$ )]. These data, in combination with fragment ions having m/z 451 ([M ] ${ }^{+}-\mathrm{Me}$ ), 391 ([M ] ${ }^{+}-\mathrm{Me}-\mathrm{HOAc}$ ), 313 [loss of side chain (s.c.; $\left.\mathrm{C}_{8} \mathrm{H}_{15}\right)-42$ ], $299\left(313-\mathrm{CH}_{2}\right)$, 295 (loss of s.c. and HOAc), 253 (295-42), and $69\left[\mathrm{CH}_{2} \mathrm{CH}=\right.$ $\left.\mathrm{C}(\mathrm{Me})_{2}\right]^{+}(\mathrm{C}-23-\mathrm{C}-27)^{5}$ suggested that $\mathbf{2 a}$ had a $3 \beta$-acetoxy tetracyclic triterpene skeleton possessing a $\Delta^{7,9(11)}$-diene system and a $\mathrm{C}_{8}$-side chain containing an isopropylidene group. The diagnostic fragment ion at $\mathrm{m} / \mathrm{z} 185\left[\mathrm{C}_{14} \mathrm{H}_{17}\right]^{+}$ formed by the loss of ring D plus side chain due to cleavage of the C-8-C-14 and C-11-C-12 bonds with concomitant HOAC loss supported the $\Delta^{7,9(11)}$-unsaturation. The above evidence coupled with the ${ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$ NMR data in addition to analysis of ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY, HMQC, and HMBC spectra indicated that 2a possesses a 4,4,14-trimethyl- $\Delta^{7,9(11), 24-}$ statrien-3 $\beta$-ol acetate structure. The stereochemistry of 2a was established by phase-sensitive NOESY experiment. Compound 2a showed significant NOE correlations between $[\mathrm{H}-29(4 \beta-\mathrm{Me})-\mathrm{H}-2 \beta$ and $\mathrm{H}-6 \beta-\mathrm{H}-19(10 \beta-\mathrm{Me})-\mathrm{H}-$ $30(14 \beta-\mathrm{Me})-\mathrm{H}-17 \beta-\mathrm{H}-21]$ on the $\beta$-face, $[\mathrm{H}-3 \alpha-\mathrm{H}-28(4 \alpha-$ $\mathrm{Me})$ and $\mathrm{H}-5 \alpha$ ] and $[\mathrm{H}-18(13 \alpha-\mathrm{Me})-\mathrm{H}-20$ ] on the $\alpha$-face, and $[\mathrm{H}-16 \alpha, \beta-\mathrm{H}-21]$ of the molecule (Figure 1). ${ }^{6}$ These NOE correlations were consistent with those for two euphane-typetriterpenes, butyrospermol (eupha-7,24-dien$3 \beta$-ol) acetate and euphol acetate (1a), ${ }^{8,9}$ thus suggesting that 2a possesses the same stereochemistry at the C/D ring junction (13, $14 \beta$ ), C-17 (17 $\beta-\mathrm{H}$ ), and C-20 (20R) as that of butyrospermol acetate and $\mathbf{1 a}$. We concluded that $\mathbf{2 a}$ is eupha-7,9(11),24-trien-3 $\beta$-ol acetate (antiquol C acetate). Alkaline hydrolysis of $\mathbf{2 a}$ yielded eupha-7,9(11),24-trien$3 \beta$-ol (2; antiquol C; $\mathrm{C}_{30} \mathrm{H}_{48} \mathrm{O} ;[\mathrm{M}]^{+} \mathrm{m} / \mathrm{z} 424.3682$ ).

Compound 3a, which showed [M] ${ }^{+}$at m/z 468.3972 $\left(\mathrm{C}_{32} \mathrm{H}_{52} \mathrm{O}_{2}\right)$ in the HREIMS, had a secondary acetoxyl group $\left[v_{\max } 1732 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 2.02(3 \mathrm{H}, \mathrm{s})\right.$ and $4.70(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=3.1$



3

7
$\mathrm{Hz})$ ], two trisubstituted double bonds $\left[v_{\max } 830 \mathrm{~cm}^{-1}\right.$; $\delta_{\mathrm{H}}$ $5.10(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz})$ and $5.56(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=5.2 \mathrm{~Hz})$ ], a terminal isopropylidene group [ $\delta_{H} 1.61(\mathrm{~s})$ and $1.69(\mathrm{~s})$ ], and five tertiary $\left[\delta_{\mathrm{H}} 0.81,0.84,0.88,1.05\right.$, and 1.08 (each s)] and one secondary [ $\delta_{\mathrm{H}} 0.86(\mathrm{~d}, \mathrm{~J}=6.1 \mathrm{~Hz})$ ] methyl groups. This, in combination with fragment ions having m/z 453 ([M] ${ }^{+}$- Me), 393 ([M] ${ }^{+}-\mathrm{Me}-\mathrm{HOAc}$ ), 355 [loss of $\mathrm{C}_{8} \mathrm{H}_{15}$ and 2 H ], and $69\left[\mathrm{CH}_{2} \mathrm{CH}=\mathrm{C}(\mathrm{Me})_{2}\right]^{+}, 5$ suggested that compound 3a was a triterpene with a tetracyclic skeleton possessing one double bond, an axially oriented acetoxyl group located most likely at $\mathrm{C}-3$, and a $\mathrm{C}_{8}$-side chain containing an isopropylidene functionality. The highly deshielded methine ${ }^{1} \mathrm{H}$ signal ( $\delta_{\mathrm{H}} 5.56$ ) due to a skeletal double bond suggested that it was located at C-5(6). ${ }^{2,8,10}$ Further, the highly deshielded ${ }^{13} \mathrm{C}$ signal at $\delta_{\mathrm{C}} 142.3$ due to $\mathrm{C}-5$ was consistent with the corresponding signal in the spectrum of 19(10 $\rightarrow 9$ )abeo-8 $\beta, 9 \beta, 10 \alpha$-Ianost-5-ene-(10 $\alpha-$ cucurbit-5-ene) $)^{2,11}$ and $19(10 \rightarrow 9)$ abeo- $8 \alpha, 9 \beta, 10 \alpha$-euph/ tirucall-5-ene skel etal structures. ${ }^{10,12}$ The diagnostic fragment ions at $\mathrm{m} / \mathrm{z} 274\left[\mathrm{C}_{20} \mathrm{H}_{34}\right]^{+}$and $134\left[\mathrm{C}_{10} \mathrm{H}_{14}\right]^{+}$formed by the cleavage of the $\mathrm{C}-7-\mathrm{C}-8$ and $\mathrm{C}-9-\mathrm{C}-10$ bonds, resulting from retro-Diels-Alder cleavage in ring $B,{ }^{13}$ ruled out the possibility of the $10 \alpha$-cucurbit-5-ene-skel etal structure. The above evidence coupled with the analysis of ${ }^{1} \mathrm{H}-$


Figure 1. Major NOE correlations $(\leftarrow \rightarrow)$ for $\mathbf{2 a}, \mathbf{3 a}$, and $\mathbf{4 a}$.
${ }^{1} \mathrm{H}$ COSY, HMQC, and HMBC spectra indicated that 3a possesses a $19(10 \rightarrow 9)$ abeo- $8 \alpha, 9 \beta, 10 \alpha$-eupha/tirucalla-5,24-diene-3 $\beta$-ol acetate structure. The stereochemistry of 3a was established by phase-sensitive NOESY experiment. Compound 3a showed significant NOE correlations between [H-1 $\beta-\mathrm{H}-19(10 \beta-\mathrm{Me})-\mathrm{H}-30(14 \beta-\mathrm{Me})-\mathrm{H}-17 \beta-\mathrm{H}-21]$ on the $\beta$-face, $[\mathrm{H}-3 \alpha-\mathrm{H}-28(4 \alpha-\mathrm{Me})-\mathrm{H}-10 \alpha-\mathrm{H}-8 \alpha-\mathrm{H}-18(13 \alpha-$ $\mathrm{Me})-\mathrm{H}-20]$ on the $\alpha$-face, and $[\mathrm{H}-16 \alpha, \beta-\mathrm{H}-21]$ of the molecule (Figure 1). ${ }^{6}$ These NOE correlations suggest that 3a possesses the same stereochemistry at the C/D ring junction, C-17, and C-20 as that of euphane-type triterpenes. ${ }^{8,9}$ We concluded that 3a is 19(10 $\rightarrow 9$ )abeo- $8 \alpha, 9 \beta, 10 \alpha-$ eupha-5,24-dien-3 $\beta$-ol acetate. Alkaline hydrolysis of 3a yielded $19(10 \rightarrow 9)$ abeo- $8 \alpha, 9 \beta, 10 \alpha$-eupha-5,24-dien-3 $\beta$-ol (3, $\mathrm{C}_{30} \mathrm{H}_{50} \mathrm{O}$ ).
Compound 4a, which exhibited [M] ${ }^{+}$at m/z 482.4120 $\left(\mathrm{C}_{33} \mathrm{H}_{54} \mathrm{O}_{2}\right)$ in the HREIMS, had a secondary acetoxyl group $\left[v_{\max } 1738 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 2.05(3 \mathrm{H}, \mathrm{s})\right.$ and $4.50(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=4.3$, 11.6 Hz )], a terminal methylene group [ $v_{\max } 891 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}$ 4.66 and 4.71 (each 1 H and s)], and five tertiary [ $\delta_{\mathrm{H}} 0.75$ $(3 \mathrm{H}), 0.87(6 \mathrm{H})$, and $0.88(6 \mathrm{H})$ (each s)] and three secondary [ $\delta_{H} 0.93,1.02$, and 1.03 (each 3 H and d)] methyl groups. This, in combination with fragment ions having m/z 467 ([M] ${ }^{+}-\mathrm{Me}$ ), 407 ([M ] ${ }^{+}-\mathrm{Me}-\mathrm{HOAC}$ ), 297 [loss of $\mathrm{C}_{9} \mathrm{H}_{17}$ and HOAC], 255 (297-42), and 241 ( $255-\mathrm{CH}_{2}$ ), ${ }^{5}$ suggested that compound 4a was a triterpene with a tetracyclic skeleton possessing a tetrasubstituted $\Delta^{8}$-double bond, an axially oriented acetoxyl group located most likely at C-3, and a $\mathrm{C}_{9}$-side chain containing a terminal methylene group. A further diagnostic ion at $\mathrm{m} / \mathrm{z} 383\left([\mathrm{M}]^{+}-\mathrm{C}_{6} \mathrm{H}_{12}-\mathrm{Me}\right)$ due to a McLafferty rearrangement involving cleavage of the C-22-C-23 bond suggested the presence of a terminal methylene group at C-24 in the side chain. ${ }^{5}$ Furthermore, the ${ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$ NMR data in addition to analysis of ${ }^{1} \mathrm{H}-$ ${ }^{1} \mathrm{H}$ COSY, HMQC, and HMBC spectra enabled the structure of 4 a to be formulated as 24-methylene-4,4,14-trimethyl- $\Delta^{8}$-sten- $3 \beta$-ol acetate with as-yet-to-be-determined stereochemistry. Compound 4a exhibited definite NOE

Table 1. ${ }^{13} \mathrm{C},{ }^{1} \mathrm{H}$, and HMBC NMR Spectral Data for Triterpene Acetates $\mathbf{2 a} \mathbf{-} \mathbf{4 a}\left(\mathrm{CDCl}_{3}\right)$

|  | 2a |  |  | 3a |  |  | 4a |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C no. | $\delta_{\text {c }}$ | $\delta_{H}{ }^{\text {a }}$ | HMBC <br> ( H to C ) | $\delta_{\text {c }}$ | $\delta_{H}{ }^{\text {a }}$ | HMBC <br> ( H to C ) | $\delta_{\text {c }}$ | $\delta_{\mathrm{H}}{ }^{\text {a }}$ | HMBC <br> ( H to C ) |
| 1 | 35.3 | 1.66 (a), 1.80 (b) | 2, 3, 5, 10, 19 | 19.5 | 1.57 (a), 1.47 (b) | 2 | 34.9 | 1.30 (a), 1.77 (b) | 2, 10, 19 |
| 2 | 24.2 | 1.76 (a), 1.69 (b) | 1, 4 | 25.5 | 1.84 (a), 1.76 (b) | 4, 5, 29 | 24.2 | 1.72 (a), 1.63 (b) | 1, 3 |
| 3 | 80.9 | 4.51 (dd, 4.4, 10.8) | $2,4,10,28,29$ | 78.6 | 4.70 (t, 3.1) | 1, 5, COMe | 80.9 | 4.50 (dd, 4.3, 11.6) | $1,2,4,28,29$ |
| 4 | 37.9 |  |  | 39.2 |  |  | 37.9 |  |  |
| 5 | 48.4 | 1.36 (dd, 4.8, 11.3) | $\begin{aligned} & 3,4,6,7,10 \\ & 19,28,29 \end{aligned}$ | 142.3 |  |  | 51.1 | 1.22 (br d, 11.9) | 1, 6, 7, 10, 29 |
| 6 | 23.5 | 2.20 (a), 2.07 (b) | 5, 7, 8, 10 | 119.8 | 5.56 (d, 5.2) | 4, 7, 8, 10 | 18.8 | 1.69 (a), 1.43 (b) | 5, 7, 10 |
| 7 | 118.1 | 5.33 (br s) | 5, 6, 14 | 25.1 | 1.65 (a), 2.08 (b) | 5, 6, 9 | 27.5 | 1.94 (a), 2.11 (b) | 6, 8, 9 |
| 8 | 141.5 |  |  | 44.5 | 1.67 (dd, 4.4, 11.4) | 7, 9, 10, 14, 19, 30 | 133.6 |  |  |
| 9 | 144.7 |  |  | 35.1 |  |  | 133.9 |  |  |
| 10 | 36.0 |  |  | 49.6 | 2.10 (br d, 11.0) | 1, 5, 9, 19 | 37.1 |  |  |
| 11 | 115.6 | 5.20 (br s) | 10, 12, 13 | 35.7 | 1.53 (2H) | 8, 9, 12, 19 | 21.5 | 2.06 (a), 1.91 (b) | 8, 9, 12 |
| 12 | 38.4 | 2.18 (2H) | 9, 11, 13, 14, 18 | 30.4 | 1.55 (a), 1.76 (b) | 11, 13, 14, 18 | 30.8 | 1.70 (2H) | $11,13,14,17,18$ |
| 13 | 44.1 |  |  | 46.1 |  |  | 44.1 |  |  |
| 14 | 49.7 |  |  | 47.5 |  |  | 49.9 |  |  |
| 15 | 31.1 | 1.64 (a), 1.31 (b) | 8, 13, 14, 17, 30 | 34.1 | 1.13 (a), 1.22 (b) | 13, 14 | 29.8 | 1.55 (a), 1.19 (b) | 13, 14, 16 |
| 16 | 28.1 | 1.95 (a), 1.34 (b) | 13, 14, 17 | 28.2 | 1.30 (a), 1.88 (b) | 13, 17 | 28.1 | 1.33 (a), 1.91 (b) | 15, 14, 17, 20 |
| 17 | 50.8 | 1.6 | $\begin{aligned} & 12,13,14, \\ & 16,20,21 \end{aligned}$ | 49.9 | 1.54 | 12, 13 | 50.1 | 1.51 | $\begin{aligned} & 12,13,14,16, \\ & 18,20 \end{aligned}$ |
| 18 | 16.2 | 0.62 (s) | 12, 13, 14, 17 | 15.4 | 0.81 (s) | 12, 13, 14, 17 | 15.4 | 0.75 (s) | 12, 13, 14, 17 |
| 19 | 20.8 | 0.94 (s) | 1, 5, 9, 10 | 16.3 | 0.84 (s) | 8, 9, 10, 11 | 20.2 | 0.88 (s) | 1, 5, 9, 10 |
| 20 | 35.7 | 1.45 | 21 | 35.2 | 1.52 | 17, 21, 22 | 36.5 | 1.42 | 17, 22 |
| 21 | 18.8 | 0.86 (d, 6.0) | 17, 20, 22 | 18.8 | 0.86 (d, 6.1) | 17, 20, 22 | 18.8 | 0.93 (d, 6.4) | 17, 20, 22, 24 |
| 22 | 35.3 | 1.05, 1.59 | 20, 21, 23 | 35.4 | 1.11, 1.58 | 21 | 35.0 | 1.11, 1.56 | 20, 23 |
| 23 | 25.0 | 1.87, 2.06 | 24, 25 | 24.7 | 1.83, 2.01 | 22, 24 | 31.3 | 1.89, 2.10 | 22, 24, 241 |
| 24 | 125.1 | 5.11 (t, 6.0) | 23, 26, 27 | 125.2 | 5.10 (t, 7.0) | 26, 27 | 156.9 |  |  |
| 25 | 131.0 |  |  | 130.9 |  |  | 33.8 | 2.23 (sept., 6.7) | 23, 24, 241, 26, 27 |
| 26 | 25.7 | 1.69 (s) | 24, 25, 27 | 25.8 | 1.69 (s) | 24, 25, 26 | 22.0 | 1.02 (d, 6.7) | 24, 25, 27 |
| 27 | 17.7 | 1.61 (s) | 24, 25, 26 | 17.7 | 1.61 (s) | 24, 25, 27 | 21.8 | 1.03 (d, 7.0) | 24, 25, 26 |
| 28 | 27.6 | 0.87 (s) | 3, 4, 5, 29 | 29.1 | 1.08 (s) | 3, 4, 5, 29 | 28.0 | 0.88 (s) | 3, 4, 5, 29 |
| 29 | 16.4 | 0.96 (s) | 3, 4, 5, 28 | 25.1 | 1.05 (s) | 3, 4, 5, 28 | 16.6 | 0.87 (s) | 3, 4, 5, 28 |
| 30 | 23.2 | 0.84 (s) | 8, 13, 14, 15 | 19.1 | 0.88 (s) | 8, 13, 14, 15, 17 | 24.4 | 0.87 (s) | 8, 13, 14, 15 |
| 241 |  |  |  |  |  |  | 106.0 | 4.66 (s), 4.71 (s) | 23, 25, 26 |
| COMe | 21.3 | 2.06 (s) | COMe | 21.3 | 2.02 (s) | COMe | 21.3 | 2.05 (s) | COMe |
| COMe | 171.0 |  |  | 170.9 |  |  | 171.0 |  |  |

${ }^{\text {a }}$ Figures in parentheses denote J values $(\mathrm{Hz})$.
correlations between $[\mathrm{H}-29(4 \beta-\mathrm{Me})-\mathrm{H}-19(10 \beta-\mathrm{Me})-\mathrm{H}-30$ ( $14 \beta-\mathrm{Me}$ ) $-\mathrm{H}-17 \beta-\mathrm{H}-21]$ on the $\beta$-face, $[\mathrm{H}-3 \alpha-\mathrm{H}-28$ ( $4 \alpha-\mathrm{Me}$ ) and $\mathrm{H}-5 \alpha$ ] and $[\mathrm{H}-18(13 \alpha-\mathrm{Me})-\mathrm{H}-20]$ on the $\alpha$-face, and [H-12-H-21] of the molecule (Figure 1), ${ }^{6}$ which were consistent with those of tirucallane-type triterpenes. ${ }^{8,9,14}$ Thus, we propose that this compound is 24-methyltirucalla-$8,24\left(24^{1}\right)$-dien- $3 \beta$-ol acetate. Alkaline hydrolysis of $\mathbf{4 a}$ yiel ded 24 -methyltirucalla-8,24(241)-dien-3 $\beta$-ol ( $4, \mathrm{C}_{31} \mathrm{H}_{52} \mathrm{O}$ ).

Table 1 shows the assigned ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data of $\mathbf{2 a}$ 4a. Signal assignments were aided by DEPT, ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY, HMQC, HMBC, and phase-sensitive NOESY experiments.

Although triterpenes with a $\Delta^{7,9(11)}$-conjugated diene system have been reported as the natural products in various skeletal types, ${ }^{4,5}$ compound $\mathbf{2}$ is the first example of a euphane-type triterpene alcohol with a $\Delta^{7,9(11) \text {-diene }}$ system as a natural product. A triterpene al cohol isolated previously from the latex of E. antiquorum, the same plant material as used in this study, was assigned the structure $10 \alpha$-cucurbita-5,24-dien-3 $\alpha$-ol (antiquol B) by Gewali et al. ${ }^{2}$ However, there was good agreement in the MS and ${ }^{1} \mathrm{H}$ NMR spectral data reported for antiquol $B^{2}$ and our compound 3 (see Experimental Section). Gewali et al. assigned the structure of antiquol $B$ mainly by NMR spectroscopy but without supporting evidence from NOE correlations. We conclude that antiquol $B$ possesses the structure $19(10 \rightarrow 9)$ abeo- $8 \alpha, 9 \beta, 10 \alpha$-eupha-5,24-dien- $3 \beta$-ol (3) but was erroneously assigned previously as having the cucurbitane-type structure. ${ }^{2}$ The structure of compound 3 has been, on the other hand, given for boeticol, a triterpene alcohol isolated from the acetone extract of Euphorbia
boetica. ${ }^{11}$ The MS and ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data reported there for boeticol and its acetate were in good agreement with those of $\mathbf{3}$ and $\mathbf{3 a}$, respectively, suggesting that antiquol $B$ and boeticol are the same compounds. ${ }^{15}$ E uphorbol, isolated from some E uphorbia species and assigned the structure 24-methyleupha-8,24(24ㄹ)-dien-3 $\beta$-ol, and its acetate ${ }^{3}$ exhibit melting points and optical rotations very close to those of $\mathbf{4}$ and $\mathbf{4 a}$ (see Experimental Section), respectively. This strongly suggests that euphorbol possesses the tirucallanetype structure but was previously assigned erroneously the euphane-type structure. ${ }^{3}$ There are fairly distinct differences in the physical characteristics between euphane- and tirucallane-type triterpenes, i.e., euphol acetate[1a; mp 117 $\left.{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}+32^{\circ}\left(\mathrm{CHCl}_{3}\right)\right]$ and tirucallol acetate [mp 163$\left.165{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}-5^{\circ}\left(\mathrm{CHCl}_{3}\right)\right],{ }^{3}$ which supports the above supposition.

The inhibitory effects of compounds $\mathbf{1 - 7}$ on EBV-EA activation induced by TPA were examined for the primary screening of antitumor-promoting activities, and the results are shown in Table 2. All compounds, with the exception of a monocyclic compound 7, showed potent inhibitory effects while preserving high viability of Raji cells. In this assay compounds 1-4 and 6 showed $100 \%$ inhibition of activation at 1000 mol ratio/TPA, of which inhibitory effects were found to be stronger than those of $\beta$-carotene, intensively studied in cancer prevention using animal models. ${ }^{16}$ The inhibitory effects against E BV-EA activation have been demonstrated to be closely parallel to those against tumor promotion in vivo. ${ }^{17,18}$ Therefore, the triterpenes 1-6 from E. antiquorum may to be useful as chemopreventive agents.

Table 2. Percentage of Epstein-Barr Virus Early Antigen Induction in the Presence of Compounds 1-7 with Respect to a Positive Control (100\%) ${ }^{a}$

|  | concentration (mol ratio/TPA) |  |  |  |  |
| :--- | :---: | :---: | :---: | :---: | :---: |
| compound | 1000 |  | 500 | 100 | 10 |
| $\mathbf{1}$ | 0 | $(70)$ | 29.8 | 75.7 | 93.5 |
| $\mathbf{2}$ | 0 | $(70)$ | 28.2 | 75.0 | 92.8 |
| $\mathbf{3}$ | 0 | $(70)$ | 34.7 | 78.4 | 96.5 |
| $\mathbf{4}$ | 0 | $(70)$ | 32.6 | 77.9 | 95.8 |
| $\mathbf{5}$ | 2.2 | $(70)$ | 33.8 | 79.4 | 97.2 |
| $\mathbf{6}$ | 0 | $(70)$ | 25.8 | 72.4 | 95.7 |
| $\mathbf{7}$ | 20.1 | $(70)$ | 51.5 | 86.3 | 100 |
| $\beta$-carotene ${ }^{\text {b }}$ | 8.6 | $(70)$ | 34.2 | 82.1 | 100 |

${ }^{\text {a }}$ Values represent relative percentages to the positive control value. TPA ( $32 \mathrm{pmol}, 20 \mathrm{ng}$ ) $=100 \%$. Values in parentheses are viability percentages of Raji cells. ${ }^{\text {b }}$ Reference compound.

## Experimental Section

General Experimental Procedures. Crystallizations were performed in MeOH , and melting points are uncorrected. Column chromatography was carried out on silica gel (Silica gel 60, Merck) and on octadecyl silica (Chromatorex-ODS, 100-200 mesh; Fuji Silysia Chemical, Ltd., Aichi, J apan). Reversed-phase HPLC was carried out on octadecyl silica columns ( $25 \mathrm{~cm} \times 10 \mathrm{~mm}$ i.d.), on a Superiorex ODS S-5 $\mu \mathrm{m}$ column (Shiseido Co., Ltd., Tokyo, J apan), and on a TSK ODS120A $5 \mu \mathrm{~m}$ (Toso Co., Tokyo, J apan), at $25^{\circ} \mathrm{C}$ with MeOH (4 $\mathrm{mL} / \mathrm{min}$ ) as mobile phase. GLC was performed using a DB-17 fused-silica capillary column ( $30 \mathrm{~m} \times 0.3 \mathrm{~mm}$ i.d., column temperature $275^{\circ} \mathrm{C}$ ). For both HPLC and GLC, cholesterol (chol est-5-en-3 $\beta$-ol) acetate was the standard for the determination of $t_{R}$ of acetoxy triterpenes. EIMS and HREIMS were recorded at 70 eV . NMR spectra were recorded at 500 MHz ( ${ }^{1} \mathrm{H}$ NMR) and $125 \mathrm{MHz}\left({ }^{13} \mathrm{C}\right.$ NMR) in $\mathrm{CDCl}_{3}$ with tetramethylsilane (TMS) ( ${ }^{1} \mathrm{H} \mathrm{NMR}$ ) and $\mathrm{CDCl}_{3}$ at $\delta 77.0\left({ }^{13} \mathrm{C} \mathrm{NMR}\right)$ as internal standard. Chemical shifts are $\delta$ values. IR and UV spectra were recorded in KBr disks and in EtOH , respectively. Specific rotations were measured at $25{ }^{\circ} \mathrm{C}$ in $\mathrm{CHCl}_{3}$. Acetylation ( $\mathrm{Ac}_{2} \mathrm{O}$-pyridine) and hydrolysis of acetates ( $5 \% \mathrm{KOH}$ in MeOH ) were performed at room temperature overnight. Triterpene acetates $\mathbf{1 a},{ }^{9,19} \mathbf{5 a},{ }^{14} \mathbf{6 a},{ }^{9}$ and $\mathbf{7 a}{ }^{20}$ were used as reference compounds.

Plant Material. The latex of Euphorbia antiquorum L. was collected in July of 1999 at Werellagama in the Central Province of Sri Lanka and identified by Mr. E. M. H. G. S. Ekanayake of the Department of Botany, University of Peradeniya, Sri Lanka. A voucher specimen (number) has been deposited at the Medical Research Institute of Sri Lanka.

Extraction and Isolation. The fresh latex ( 1250 g ) was mixed with distilled water ( 3 L ) and extracted with n -hexane ( $3 \mathrm{~L} \times 3$ ). The $n$-hexane-insol ublefraction was again extracted with EtOAc ( $3 \mathrm{~L} \times 3$ ). Evaporation of the EtOAc-soluble fraction under reduced pressure yielded EtOAc extract (120 g). A portion ( 60 g ) of the EtOAc extract was subjected to chromatography on silica gel using n-hexane-EtOAc (1:0 $\rightarrow$ $0: 1, \mathrm{v} / \mathrm{v}$ ) and MeOH as eluants. The n-hexane-EtOAc (9:1) eluted a triterpene alcohol fraction ( 42.5 g ). A portion ( 10 g ) of the fraction was acetylated, and the acetate fraction was crystallized from MeOH , yiel ding $\mathbf{1 a}(3.0 \mathrm{~g})$ and a filtrate (3.6 g). Chromatography of the latter on an octadecyl silica column, MeOH , followed by reversed-phase HPLC eventually yielded seven triterpene acetates, 1a ( 2.25 g ), 2a ( 2 mg ), 3a ( 4 mg ), $4 \mathbf{a}(23 \mathrm{mg}), 5 \mathbf{a}(4 \mathrm{mg}), \mathbf{6 a}(2 \mathrm{mg})$, and $7 \mathbf{7 a}(2 \mathrm{mg})$. On alkaline hydrolysis, these triterpene acetates yielded the corresponding free alcohols. Some physical characteristics and the spectral data of three compounds, 2-4, and their acetyl derivatives are shown below.

Antiquol C (2): amorphous gum; $[\alpha]^{25} \mathrm{D}-35.0^{\circ}$ (c 0.24); UV $\lambda_{\text {max }} 230(\log \epsilon 4.08), 237$ (4.10), 246 (3.91) nm; IR $\nu_{\text {max }} 3404$ (OH), 1639 (conjugated $-\mathrm{CH}=\mathrm{CH}-$ ), 816 ( $>\mathrm{C}=\mathrm{CH}-$ ) $\mathrm{cm}^{-1} ;{ }^{13} \mathrm{C}$ and ${ }^{1}$ H NMR C-1 $\left[\delta_{\mathrm{C}} 35.7 ; \delta_{\mathrm{H}} 1.56(\alpha), 1.82(\beta)\right]$, C-2 [27.8; 1.73 $(\alpha), 1.66(\beta)], \mathrm{C}-3$ [79.1; 3.24 (dd, J = 4.4, 11.0 Hz)], C-4 [39.0],

C-5 [48.3; 1.28 (dd, J = 4.8, 11.4 Hz)], C-6 [23.7; 2.24 ( $\alpha$ ), 2.09 ( $\beta$ )], C-7 [118.5; 5.34 (dd, J $=2.1,3.1 \mathrm{~Hz}$ )], C-8 [141.4], C-9 [145.0], C-10 [36.2], C-11 [115.4; 5.19 (t, J $=3.5 \mathrm{~Hz})$ ], C-12 [38.4; $2.17(2 \mathrm{H})$ ], C-13 [44.1], C-14 [49.7], C-15 [31.2; 1.65 ( $\alpha$ ), $1.31(\beta)]$, C-16 [28.1; 1.96 ( $\alpha$ ), 1.35 ( $\beta$ )], C-17 [50.8; 1.62]; C-18 [16.3; 0.62 (s)], C-19 [20.7; 0.92 (s)], C-20 [35.8; 1.45], C-21 [18.8; 0.87 (d, J = 7.0 Hz )], C-22 [35.3; 1.06, 1.58], C-23 [25.1; $1.87,2.07], \mathrm{C}-24$ [125.1; 5.10 (tt, J $=1.5,6.6 \mathrm{~Hz}$ )], C-25 [131.0], C-26 [25.7; 1.68 (s)], C-27 [17.7; 1.61 (s)], C-28 [27.7; 0.99 (s)], C-29 [15.3; 0.89 (s)], C-30 [23.2; 0.84 (s)]; EIMS m/z 424 [M ] ${ }^{+}$ (100), 409 (17), 391 (14), 339 (5), 337 (5), 311 (70), 295 (5), 284 (3), 271 (27), 257 (9), 253 (16), 246 (14), 240 (13), 227 (13), 213 (10), 199 (11), 185 (16), 171 (21), 69 (42); HREIMS m/z 424.3682 (calcd for $\mathrm{C}_{30} \mathrm{H}_{48} \mathrm{O}, 424.3702$ ).

Antiquol C acetate (2a): amorphous gum; $[\alpha]^{25} \mathrm{D}-23.6^{\circ}$ (c 0.33); UV $\lambda_{\max } 230$ (log $\epsilon 4.07$ ), 237 (4.09), 240 (3.99) nm; IR $v_{\max } 1738$ (OAc), 1650 (conjugated $-\mathrm{CH}=\mathrm{CH}-$ ), 816 ( $>\mathrm{C}=$ $\mathrm{CH}-) \mathrm{cm}^{-1} ; \mathrm{t}_{\mathrm{R}} 0.62$ (HPLC I), 0.36 (HPLC II), 1.34 (GLC); ${ }^{13} \mathrm{C}$ NMR and ${ }^{1}$ H NMR, seeTable 1; EIMS m/z 466 [M ] ${ }^{+}$(50), 451 (7), 406 (3), 391 (13), 381 (2), 353 (40), 337 (4), 313 (9), 295 (5), 288 (10), 273 (3), 253 (19), 240 (12), 227 (9), 213 (8), 199 (6), 185 (10), 171 (13), 69 (91), 43 (100); HREIMS m/ s 466.3842 (calcd for $\mathrm{C}_{32} \mathrm{H}_{50} \mathrm{O}_{2}, 466.3809$ ).

Antiquol B (3): needles, mp $75-76^{\circ} \mathrm{C} ;[\alpha]^{25} \mathrm{D}+24.8$ (c 0.21) [antiquol B, lit. ${ }^{2}[\alpha]^{25} \mathrm{D}+13.0^{\circ}\left(\mathrm{CHCl}_{3}\right)$; boeticol, lit. ${ }^{10}[\alpha]^{20} \mathrm{D}+$ $\left.30^{\circ}\left(\mathrm{CHCl}_{3}\right)\right]$; IR $\nu_{\max } 3434(\mathrm{OH}), 829(>\mathrm{C}=\mathrm{CH}-) \mathrm{cm}^{-1} ;{ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$ NMR: C-1 [ $\delta_{\mathrm{C}} 18.8 ; \delta_{\mathrm{H}} 1.56(\alpha)$, $1.80(\beta)$ ], C-2 [27.9; 1.87 $(\alpha), 1.70(\beta)], \mathrm{C}-3[76.4 ; 3.47(\mathrm{t}, \mathrm{J}=3.0 \mathrm{~Hz})$ ], C-4 [41.0], C-5 [142.0], C-6 [121.8; $5.64(\mathrm{~d}, \mathrm{~J}=5.8 \mathrm{~Hz})$ ], C-7 [25.2; $1.66(\alpha)$, $2.07(\beta)$ ], C-8 [44.5; 1.66], C-9 [35.1], C-10 [49.4; 2.06], C-11 [35.7; $1.55(2 \mathrm{H})$ ], C-12 [30.3; $1.55(\alpha), 1.75(\beta)$ ], C-13 [46.1], C-14 [47.4], C-15 [34.2; $1.10(\alpha), 1.21$ ( $\beta$ )], C-16 [28.2; $1.32(\alpha)$, $1.87(\beta)$ ], C-17 [49.9; 1.55]; C-18 [15.4; 0.81 (s)], C-19 [16.5; 0.84 (s)], C-20 [35.2; 1.53], C-21 [19.0; 0.86 (d, J $=6.0 \mathrm{~Hz}$ )], C-22 [35.4; 1.12, 1.57], C-23 [24.6; 1.85, 2.02], C-24 [125.2; 5.10 ( $\mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}$ )], C-25 [130.1], C-26 [25.8; 1.69 (s)], C-27 [17.7; 1.61 (s)], C-28 [29.0; 1.06 (s)], C-29 [25.5; 1.14 (s)], C-30 [18.9; 0.88 (s)]; EIMS m/z 426 [M ]+ (44), 411 (19), 408 (16), 393 (9), 313 (71), 295 (23), 274 (100), 259 (97), 231 (18), 189 (28), 173 (23), 163 (58), 134 (93); HREIMS m/z 426.3840 (calcd for $\mathrm{C}_{30} \mathrm{H}_{50} \mathrm{O}, 426.3859$ ).

Antiquol B acetate (3a): needles, mp $76-77^{\circ} \mathrm{C} ;[\alpha]^{25} \mathrm{D}$ +59.1 (c 0.22); IR $v_{\max } 1732$ (OAc), $830(>\mathrm{C}=\mathrm{CH}-)_{\mathrm{cm}^{-1} \text {; } \mathrm{t}_{\mathrm{R}}}$ 0.62 (HPLC I), 0.35 (HPLC II), 1.52 (GLC); ${ }^{13} \mathrm{C}$ NMR and ${ }^{1} \mathrm{H}$ NMR, see Table 1; EIMS m/z 468 [M ] ${ }^{+}$(7), 453 (2), 408 (9), 393 (7), 355 (29), 295 (11), 274 (40), 259 (45), 245 (6), 189 (13), 175 (20), 134 (49), 69 (93); HREIMS m/ s 468.3972 (calcd for $\mathrm{C}_{32} \mathrm{H}_{52} \mathrm{O}_{2}, 468.3964$ ).
Euphorbol (4): needles, mp 123-126 ${ }^{\circ} \mathrm{C}$ (lit. ${ }^{3} \mathrm{mp} 127-128$ ${ }^{\circ} \mathrm{C}$ ); $[\alpha]^{25} \mathrm{D}-1.0^{\circ}$ (c 0.20 ) $\left[\mathrm{lit} .{ }^{3}[\alpha]_{D} \pm 0^{\circ}\left(\mathrm{CHCl}_{3}\right)\right]$; IR $v_{\text {max }} 3082$ $(\mathrm{OH}), 889(>\mathrm{C}=\mathrm{CH}-) \mathrm{cm}^{-1} ;{ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$ NMR C-1 $\left[\delta_{\mathrm{C}} 35.3 ; \delta_{\mathrm{H}}\right.$ $1.21(\alpha), 1.77(\beta)]$, C-2 [27.9; $1.67(\alpha), 1.57(\beta)]$, C-3 [79.0; 3.24 (dd, J = 4.3, 11.6 Hz)], C-4 [38.9], C-5 [51.0; 1.12 (dd, J = 1.8, $12.5 \mathrm{~Hz})$ ], C-6 [18.9; $1.69(\alpha), 1.42(\beta)$ ], C-7 [27.7; $2.08(2 \mathrm{H})$ ], C-8 [133.5], C-9 [134.1], C-10 [37.3], C-11 [21.5; 2.06 ( $\alpha$ ), 1.92 ( $\beta$ )], C-12 [30.8; 1.72 (2H)], C-13 [44.1], C-14 [50.0], C-15 [29.8; $1.52(\alpha), 1.18(\beta)$ ], C-16 [28.1; $1.32(\alpha), 1.94(\beta)], \mathrm{C}-17[50.1 ;$ 1.52]; C-18 [15.4; 0.76 (s)], C-19 [20.2; 0.96 (s)], C-20 [36.3; 1.43], C-21 [18.8; 0.93 (d, J = 6.4 Hz )], C-22 [35.0; 1.12, 1.55], C-23 [31.3; 1.88, 2.12], C-24 [156.9], C-25 [33.8; 2.24 (sept., J $=6.7 \mathrm{~Hz})], \mathrm{C}-26$ [22.0; $1.02(\mathrm{~d}, \mathrm{~J}=6.8 \mathrm{~Hz})], \mathrm{C}-27$ [21.9; 1.03 (d, J $=7.0 \mathrm{~Hz}$ )], C-28 [28.0; 1.00 (s)], C-29 [15.5; 0.80 (s)], C-30 [24.4; $0.88(\mathrm{~s})], \mathrm{C}-24^{1}$ [105.9; 4.66 and 4.72 (each br s)]; EIMS $\mathrm{m} / \mathrm{z} 440[\mathrm{M}]^{+}$(53), 425 (100), 407 (43), 341 (11), 323 (10), 259 (33), 241 (18), 185 (18), 175 (14), 163 (22), 149 (34), 134 (55); HREIMS m/s 440.4009 (calcd for $\mathrm{C}_{31} \mathrm{H}_{52} \mathrm{O}, 440.4018$ ).
Euphorbol acetate (4a): needles, $\mathrm{mp} 125-127^{\circ} \mathrm{C}$ (lit. ${ }^{3}$ $\left.\mathrm{mp} 124-125{ }^{\circ} \mathrm{C}\right) ;[\alpha]^{25} \mathrm{D} \pm 0.0^{\circ}$ (c 0.21) $\left[\right.$ lit. $\left.{ }^{3}[\alpha]_{\mathrm{D}} \pm 0^{\circ}\left(\mathrm{CHCl}_{3}\right)\right]$; IR $v_{\max } 1738$ (OAC), 891 ( $>\mathrm{C}=\mathrm{CH}-$ ) $\mathrm{cm}^{-1}$; $\mathrm{t}_{\mathrm{R}} 0.84$ (HPLC I), 0.56 (HPLC II), 1.58 (GLC); ${ }^{13} \mathrm{C}$ NMR and ${ }^{1} \mathrm{H}$ NMR, see Table 1; EIMS m/z 482 [M ]+ (32), 467 (73), 407 (40), 383 (4), 369 (3), 323 (9), 301 (15), 283 (7), 255 (8), 247 (3), 241 (17), 229 (11), 215 (12), 189 (16), 69 (96), 43 (100); HREIMS m/ s 482.4120 (calcd for $\mathrm{C}_{33} \mathrm{H}_{54} \mathrm{O}_{2}, 482.4120$ ).

Method of EBV-EA Induction Tests. The inhibition of EBV-EA activation was assayed using Raji cells (virus nonproducer type), the EBV genome-carrying human lymphoblastoid cells, which were cultivated in $10 \%$ fetal bovine serum, Roswell Park Memorial Institute (FBS RPMI) 1640 medium sol ution (Nacalai Tesque, Inc., K yoto, J apan). The indicator cells (Raji) $\left(1 \times 10^{6} / \mathrm{mL}\right)$ were incubated at $37^{\circ} \mathrm{C}$ for 48 h in 1 mL of the medium containing n -butyric acid ( 4 mM , inducer) and 32 pmol of TPA ( $20 \mathrm{ng} / \mathrm{mL}$ ) in dimethyl sulfoxide (DMSO) and a known amount of test compound in DMSO. Smears were made from the cell suspension. The activated cells were stained by high-titer EBV-EA-positive sera from nasopharyngeal carcinoma patients and were detected by a conventional indirect immunofluorescence technique. In each assay, at least 500 cells were counted, and the experiments were repeated twice. The average EA induction was compared with that of positive control experiments with n-butyric acid plus TPA, in which EA induction was ordinarily around $30 \%$.

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